## **Preliminary communication**

## **ARENE-OLEFIN COMPLEXES OF ZEROVALENT IRON AND RUTHENIUM**

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#### Summary

Convenient syntheses of arene-olefin complexes of iron(0) and ruthenium(0), e.g.  $M(\eta^6-C_6Me_6)(\eta^4-diene)$  (M = Fe, Ru, diene = 1,3-cyclohexadiene or 1,5-cyclooctadiene),  $Fe(\eta^6-C_6Me_6)(\eta^4-C_7H_8)$  (C<sub>7</sub>H<sub>8</sub> = 1,3,5-cycloheptatriene) and  $Ru(\eta^6-C_6Me_6)(C_2H_4)_2$  are given, and some protonation reactions are described.

Arene-olefin complexes of zerovalent iron and ruthenium are of interest as potential catalysts for olefin oligomerization and for comparison with the wellknown tricarbonyl diene complexes of these elements, but few convenient synthetic routes are available. The 1,3-cyclohexadiene complexes  $M(C_6H_6)(C_6H_8)$ (M = Fe, Ru, Os) [1--3] and the *tetrahapto*-hexamethylbenzene complex  $Ru(\eta^6-C_6Me_6)(\eta^4-C_6Me_6)$  [4] have been known for some time and, more recently, the 1,5-cyclooctadiene complex  $Ru(C_6H_6)(C_8H_{12})$  has been isolated in poor yield from the reaction of  $[RuCl_2(C_6H_6)]_2$  with the diene in the presence of isopropyl-Grignard reagent [5]. Other complexes of the type  $Ru(arene)(C_8H_{12})$  are formed by displacement of 1,3,5-cyclooctatriene from  $Ru(1,3,5-C_8H_{10})(C_8H_{12})$  [6], which is itself readily accessible from the reaction of " $RuCl_3 \cdot 3H_2O$ " with 1,5cyclooctadiene and zinc dust in ethanol [7].

We have shown [8] that tetrahapto-cyclooctatetraene complexes Ru(arene)-(1-4- $\eta$ -C<sub>8</sub>H<sub>8</sub>) are readily prepared by reaction of cyclooctatetraene dianion with the readily available [RuCl<sub>2</sub>(arene)]<sub>2</sub> complexes (arene = benzene, mesitylene or hexamethylbenzene) [9-11], and we now report that the latter react with 1,5-cyclooctadiene or 1,3-cyclohexadiene in the presence of ethanol and either base (Na<sub>2</sub>CO<sub>3</sub>) or zinc dust to give yellow, air-sensitive complexes Ru(arene)-(diene) in yields of ca. 60%. The 1,3-cyclohexadiene complex Ru(C<sub>6</sub>Me<sub>6</sub>)(C<sub>6</sub>H<sub>8</sub>)\* is also isolated from the same reaction using 1,4-cyclohexadiene. The reaction between [RuCl<sub>2</sub>(C<sub>6</sub>Me<sub>6</sub>)]<sub>2</sub>, cycloheptatriene and sodium carbonate does not give

<sup>\*</sup>Except where stated otherwise, satisfactory elemental analyses and spectroscopic data have been obtained for all new compounds described herein.

the zerovalent complex  $\operatorname{Ru}(C_6\operatorname{Me}_6)(1-4-\eta-C_7\operatorname{H}_8)$  but its protonation product  $[\operatorname{Ru}(C_6\operatorname{Me}_6)(1-5-\eta-C_7\operatorname{H}_9)]^+$ , isolated as its  $\operatorname{PF}_6^-$  or  $\operatorname{BF}_4^-$  salt. These reactions reveal the close analogy between  $[\operatorname{Ru}\operatorname{Cl}_2(\operatorname{arene})]_2$  and the isoelectronic pentamethylcyclopentadienylrhodium(III) complex  $[\operatorname{Rh}\operatorname{Cl}_2(\eta^5-C_5\operatorname{Me}_5)]_2$  [12-14]. Reaction between  $[\operatorname{Ru}\operatorname{Cl}_2(\operatorname{C}_6\operatorname{Me}_6)]_2$ , aqueous ethanolic  $\operatorname{Na}_2\operatorname{CO}_3$  and ethylene gives the tan, air-sensitive ruthenium(0) bis(ethylene) complex  $\operatorname{Ru}(\operatorname{C}_6\operatorname{Me}_6)(\operatorname{C}_2\operatorname{H}_4)_2$  (<sup>1</sup>H NMR,  $\operatorname{C}_6\operatorname{D}_6$ , 32°C:  $\delta$  (ppm) 1.70 (s, 18H,  $\operatorname{C}_6\operatorname{Me}_6$ ), 1.48 (m, 4H,  $\operatorname{C}_2\operatorname{H}_4$ ), 1.04 (m, 4H,  $\operatorname{C}_2\operatorname{H}_4$ )] in ca. 40% yield. Although the ethylene resonances are broadened considerably at 116°C in  $\operatorname{C}_6\operatorname{F}_5\operatorname{B}$ r, complete coalescence is not observed, indicating a higher barrier to ethylene rotation than in the analogous isoelectronic rhodium(I) complexes  $\operatorname{Rh}(\operatorname{C}_5\operatorname{R}_5)(\operatorname{C}_2\operatorname{H}_4)_2$  ( $\operatorname{R} = \operatorname{H}, \operatorname{CH}_3$ ) [12, 15]. Some related iron(0) complexes have been obtained by reduction of salts of

 $[Fe(arene)_2]^{2^+}$  with cyclooctatetraene dianion  $C_8H_8^{2^-}$  or with sodium amalgam in tetrahydrofuran. Thus, addition of either of these reducing agents to  $[Fe(C_6Me_6)_2](PF_6)_2$  gives a brown-black solution of bis(hexamethylbenzene)iron(0),  $Fe(C_6Me_6)_2$  [16]\*, which on reaction with cycloheptatriene at 40°C for 16 h, gives the orange complex  $Fe(C_6Me_6)(1-4-\eta-C_7H_8)$  in 51% yield. Under similar conditions both 1,3- and 1,4-cyclohexadiene gives the 1,3-cyclohexadiene complex  $Fe(C_6Me_6)(C_6H_8)$ , and 1,5-cyclooctadiene gives a not fully characterised, thermally unstable orange-red complex which is probably  $Fe(C_6Me_6)(1,5-C_8H_{12})$ .

The new complexes show the expected reactivity towards electrophiles. Protonation (HBF<sub>4</sub> or HPF<sub>6</sub>) at the free double bond of Fe(C<sub>6</sub>Me<sub>6</sub>)(C<sub>7</sub>H<sub>8</sub>) gives the 1—5- $\eta$ -cycloheptadienyl complex [Fe(C<sub>6</sub>Me<sub>6</sub>)(1—5- $\eta$ -C<sub>7</sub>H<sub>9</sub>)]<sup>+</sup> and addition of Ph<sub>3</sub>C<sup>+</sup> BF<sub>4</sub> gives the corresponding trityl-substituted cation [Fe(C<sub>6</sub>Me<sub>6</sub>)(1—5- $\eta$ -C<sub>7</sub>H<sub>8</sub>CPh<sub>3</sub>)]<sup>+</sup> without abstraction of hydride ion; similar reactions of cycloheptatrieneiron tricarbonyl are well known [18, 19]. Protonation (HPF<sub>6</sub>) of the 1,3-cyclohexadiene complex Ru(C<sub>6</sub>Me<sub>6</sub>)(C<sub>6</sub>H<sub>8</sub>) gives a salt which, in the solid state, appears to be the coordinately unsaturated, 16-electron 1—3- $\eta$ -cyclohexenyl species [Ru(C<sub>6</sub>Me<sub>6</sub>)(1—3- $\eta$ -C<sub>6</sub>H<sub>9</sub>)]PF<sub>6</sub>. However, the <sup>1</sup>H NMR spectrum (CD<sub>2</sub>Cl<sub>2</sub>, 32°C:  $\delta$  (ppm) 2.30 (s, 18H, C<sub>6</sub>Me<sub>6</sub>), 2.97 (approx. q, 6H, *exo*-CH<sub>2</sub>), -2.90 (approx. spt, 3H, *endo*-CH<sub>2</sub> + RuH)) indicates that *endo*-methylene hydrogen atoms migrate rapidly on the NMR time-scale between carbon and ruthenium (eq. 1).

At  $-80^{\circ}$ C the signals at  $\delta$  2.97 and -2.90 ppm have almost disappeared but a limiting spectrum could not be reached. This behaviour is entirely analogous to that reported for the isoelectronic complex  $[Rh(\eta^5-C_5H_5)(1-3-\eta-C_6H_9)]^+$  generated by protonation of  $Rh(C_5H_5)(C_6H_8)$  [20]. Addition of CO gives the coordinately saturated complex  $[Ru(C_6Me_6)(1-3-\eta-C_6H_9)(CO)]^+$  which exhibits the expected <sup>1</sup>H NMR spectrum for a static  $\eta^3$ -allyl complex.

We are currently extending the scope of these preparative routes to other olefins and are investigating electrophilic and nucleophilic reactions of the resulting complexes.

\*During the course of our work the preparation of  $Fe(C_6Me_6)_2$  by sodium naphthalene reduction of  $[Fe(C_6Me_6)_2](PF_6)_2$  and the reaction of  $Fe(C_6Me_6)_2$  with 1.3-cyclohexadiene were reported independently [17].

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